

AMENDMENTS TO THE CLAIMS:

This listing of claims will replace all prior versions, and listings, of claims in the application:

Listing of Claims:

1. (Original) A lithium-nickel-cobalt-manganese-containing composite oxide represented by a general formula, $\text{Li}_p\text{Ni}_x\text{Mn}_{1-x-y}\text{Co}_y\text{O}_2\text{-qF}_q$ (where $0.98 \leq p \leq 1.07$, $0.3 \leq x \leq 0.5$, $0.1 \leq y \leq 0.38$, and $0 \leq q \leq 0.05$), formed by synthesizing coagulated particles of a nickel-cobalt-manganese composite hydroxide wherein primary particles obtained by precipitating the nickel-cobalt-manganese composite hydroxide are coagulated to form secondary particles, by supplying an aqueous solution of a nickel-cobalt-manganese salt, an aqueous solution of an alkali-metal hydroxide and an ammonium-ion donor continuously or intermittently to a reaction system, and making the reaction proceed in the state wherein the temperature of said reaction system is substantially constant within a range between 30 and 70°C, and pH is maintained at a substantially constant value within a range between 10 and 13; synthesizing coagulated particles of a nickel-cobalt-manganese composite oxyhydroxide by making an oxidant act on said coagulated composite hydroxide particles; and dry-blending at least said composite oxyhydroxide and a lithium salt, and firing the mixture in an oxygen-containing atmosphere.
2. (Original) The lithium-nickel-cobalt-manganese-containing composite oxide according to claim 1, characterized in that the density of the compressed powder is 2.6 g/cm² or more.
3. (Currently Amended) The lithium-nickel-cobalt-manganese-containing composite oxide according to claim 1-~~or 2~~, characterized in having an R-3m rhombohedral structure.
4. (Currently Amended) A method for manufacturing a lithium-nickel-cobalt-manganese-containing composite oxide represented by a general formula, $\text{Li}_p\text{Ni}_x\text{Mn}_{1-x-y}\text{Co}_y\text{O}_2\text{-qF}_q$ (where $0.98 \leq p \leq 1.07$, $0.3 \leq x \leq 0.5$, $0.1 \leq y \leq 0.38$, and $0 \leq q \leq 0.05$)

according to claim 1 ~~any one of claims 1 to 3~~, comprising:

a step for synthesizing coagulated particles of a nickel-cobalt-manganese composite hydroxide wherein primary particles obtained by precipitating the nickel-cobalt-manganese composite hydroxide are coagulated to form secondary particles, by supplying an aqueous solution of a nickel-cobalt-manganese salt, an aqueous solution of an alkali-metal hydroxide and an ammonium-ion donor continuously or intermittently to a reaction system, and making the reaction proceed in the state wherein the temperature of said reaction system is substantially constant within a range between 30 and 70°C, and pH is maintained at a substantially constant value within a range between 10 and 13;

a step for synthesizing coagulated particles of a nickel-cobalt-manganese composite oxyhydroxide by making an oxidant act on said coagulated composite hydroxide particles; and

a step for dry-blending at least said coagulated composite oxyhydroxide particles and a lithium salt, and firing the mixture in an oxygen-containing atmosphere.

5. (Original) The method for manufacturing a lithium-nickel-cobalt-manganese-containing composite oxide according to claim 4, wherein the lithium salt is lithium carbonate.

6. (Original) A material for a positive electrode active material for a lithium secondary cell consisting of coagulated particles of a nickel-cobalt-manganese composite oxyhydroxide represented by a general formula, $Ni_xMn_{1-x-y}Co_yOOH$ (where $0.3 \leq x \leq 0.5$, and $0.1 \leq y \leq 0.38$), formed by synthesizing coagulated particles of a nickel-cobalt-manganese composite hydroxide wherein primary particles obtained by precipitating the nickel-cobalt-manganese composite hydroxide are coagulated to form secondary particles, by supplying an aqueous solution of a nickel-cobalt-manganese salt, an aqueous solution of an alkali-metal hydroxide and an ammonium-ion donor continuously or intermittently to a reaction system, and making the reaction proceed in the state wherein the temperature of said reaction system is substantially constant within a range between 30 and 70°C, and pH is maintained at a substantially constant value within a range between 10 and 13; and making an oxidant act on said coagulated composite hydroxide particles.

7. (Original) The material for a positive electrode active material for a lithium secondary cell according to claim 6, characterized in that the specific surface area is 4 to 30 m²/g.

8. (Currently Amended) The material for a positive electrode active material for a lithium secondary cell according to claim 6-~~or~~-7, characterized in that the density of the compressed powder is 2.0 g/cm² or more.

9. (Currently Amended) The material for a positive electrode active material for a lithium secondary cell according to claim 6,~~7 or 8~~, characterized in that the half-value width of the diffraction peak when 2Θ is $19 \pm 1^\circ$ in X-ray diffraction using Cu-K α lines is 0.3 to 0.5 $^\circ$.

10. (Currently Amended) A method for manufacturing the material for a positive electrode active material for a lithium secondary cell represented by a general formula, $Ni_xMn_{1-x}Co_yOOH$ (where $0.3 \leq x \leq 0.5$, and $0.1 \leq y \leq 0.38$), according to claim 6~~any one of claims 6 to 9~~, comprising:

a step for synthesizing coagulated particles of a nickel-cobalt-manganese composite hydroxide wherein primary particles obtained by precipitating the nickel-cobalt-manganese composite hydroxide are coagulated to form secondary particles, by supplying an aqueous solution of a nickel-cobalt-manganese salt, an aqueous solution of an alkali-metal hydroxide and an ammonium-ion donor continuously or intermittently to a reaction system, and making the reaction proceed in the state wherein the temperature of said reaction system is substantially constant within a range between 30 and 70 $^\circ$ C, and pH is maintained at a substantially constant value within a range between 10 and 13; and

a step for synthesizing coagulated particles of a nickel-cobalt-manganese composite oxyhydroxide by making an oxidant act on said coagulated composite hydroxide particles.